

Chem 161L - General Chemistry I Laboratory 1 credit (CRN 61273) Monday 8:30 – 11:15 PM, Imiloa 131

INSTRUCTOR: Marc R. Bresler

OFFICE: Imiloa 112B or via zoom (a) https://hawaii.zoom.us/j/5803716362

OFFICE HOURS: Monday 7:30-8:30AM or by appointment

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WINDWARD COMMUNITY COLLEGE MISSION STATEMENT

Windward Community College offers innovative programs in the arts and sciences and opportunities to gain knowledge and understanding of Hawai'i and its unique heritage. With a special commitment to support the access and educational needs of Native Hawaiians, we provide the Ko'olau region of O'ahu and beyond with liberal arts, career and lifelong learning in a supportive and challenging environment — inspiring students to excellence.

CATALOG DESCRIPTION

CHEM 161L is a laboratory course that accompanies CHEM 161, the first course of a two-course sequence designed to meet the one-year requirement of general college chemistry. Experiments are performed which relate to the lecture material in CHEM 161. The student will develop practical laboratory skills to competently and safely use laboratory equipment.

Prerequisite: MATH 103 with a grade of C or better.

Corequisite: Credit or concurrent registration in CHEM 161

Rec Prep: CHEM 151 or High School Chemistry

ACTIVITIES REQUIRED OUTSIDE CLASS TIMES

On average, you should spend about **3-4 hours per week outside** the classroom to study for this course.

- 1-2 hours per week to read the lab activity and complete the Pre-Lab assignment.
- 1-2 hours per week completing the report. You are expected to meet the deadline for submission. To reduce the time involved, you should do the calculations and Post-Lab analysis in class immediately after performing the lab while it is fresh in your mind. A formal lab report will take much longer to complete.

STUDENT LEARNING OUTCOMES

Perform laboratory experiments in a safe and competent manner with appropriate safety considerations and actions.

Perform laboratory experiments to illustrate and verify chemistry laws.

Perform laboratory experiments to determine the identity, composition and/or structure of an unknown compound.

Show critical thinking by synthesizing conclusions based on observations and data.

Make and record precise measurements, calculate results using significant figures, standard deviations and identify sources of error in laboratory experiments.

Express in writing the results of experiments by preparing lab reports using the standard scientific format.

PURPOSE OF THE LABORATORY COURSE

The chemistry laboratory allows the student to understand some of the theories discussed in the lecture more thoroughly. In the laboratory, you will be involved with the processes of scientific inquiry used to discover chemical principles. It is the only way for the student to learn the techniques that are so important in research and in most laboratories. The student will discover that doing quality work in the laboratory requires a great deal of patience and care.

LEARNING RESOURCES

- 1) Lab Manual (purchased at bookstore or DL offline Laulima)
- 2) Website: http://laulima.hawaii.edu (use UH email login and password)
- 3) Scientific Calculator: (Recommend TI -30XIIS from CVS or Walmart for about 15\$)
- 4) Goggles (for purchase ~10\$ at bookstore or hardware type store)

COURSE TASKS AND GRADING

- Pre-Lab Assignments. Read through the introduction and experimental procedure, and complete the online Pre-Lab assignment before the start of class. Be sure to submit before deadline, because late submissions are not accepted. Know the objectives and figure out beforehand exactly what you are going to do in the lab. The prelab must be signed before initiating the experiment.
- Attendance and hands-on activity. Attendance in Pre-Lab Discussion (in Imiloa 111) and actual performance of experiment in the laboratory (in Imiloa 131) are required. You will record observation/data, calculate/analyze results and answer post-lab questions in the lab. You must get the professor's signature after completing the lab.
- Laboratory reports. You are expected to turn in an *individual* report for each lab experiment at the start of the next meeting. The format of each lab is described in the course schedule. Include all data and calculations, and answer ALL questions. Use internet resources and Lecture Notes/textbook to answer post-lab questions.
- One Exam. Exam will take about **1.5 hours**. This will be closed note unless otherwise stated. A copy of the Periodic Table and physical constants will be provided.

The final grade will be based on the following scheme:

13 Pre-Lab Assignments, 10 points each	130 points
12 Lab Worksheets, 25 points each	300 points
2 Formal Lab Report (Draft and Final)50 points each	100 points
Lab Final	100 Points
Total Points	630 points

Course grades will be assigned as follows:

A-- 90-100% of the cumulative points possible

B-- 80-89% of the cumulative points possible

C-- 70-79% of the cumulative points possible

D-- 60-69% of the cumulative points possible

F-- less than 60% of the cumulative points possible

For grades I, W, Cr, NC-- See college catalog, https://windward.hawaii.edu/Catalogs_Schedules/WCC_Catalog_current.pdf

The deadline to change from A-F to CR/NC/audit option (with Office of Admissions & Records) is on October 31, 2023.

POLICIES

- 1. Do NOT miss a laboratory. There are no make-ups available.
- 2. Laboratory safety rules must be followed. Lab coats (or long-sleeve shirts), goggles, long pants and close-toed shoes are required at all times inside the laboratory. If you do not have the proper clothing, you may not be able to work in the lab.
- 3. The use of personal mobile devices such as cell phones, iPods, video game players, etc. is strictly prohibited during lecture, quizzes, and exams. The use of any such device during a quiz or exam will be treated as *academic dishonesty*.
- 4. Plagiarism in any form is considered *academic dishonesty*.
- 5. *Academic dishonesty will not be tolerated* and may result in judicial proceedings in accordance with UH policy.
- 6. Any student exhibiting *disruptive behavior* as defined by the Student Conduct

Code at Windward CC may be asked by the instructor to leave the class immediately.

LEARNING RESOURCES

Required: Chemistry 161L Laboratory Manual, edited by L U Colmenares
Optional: Electronic Lab Notebook https://www.labarchives.com/

Course Website: http://laulima.hawaii.edu (use UH email account login and password)

Required Textbook: https://openstax.org/details/books/chemistry (free download)

Other Requirements: Scientific Calculator (phone not permitted during exam), Internet Access, Lab goggles,

closed toed shoes and a lab gown if you wear short pants/skirt/dress or low-waist

pants/skirts.

Online Tutoring: Tutor.com online tutoring open 24/7 https://windward.hawaii.edu/tutor.com/

ADDITIONAL INFORMATION

ATTENDANCE and PRE-LAB POLICY. Attendance will be taken at the start of every class meeting. Important background information about the experiment and safety will be discussed in the **Pre-Lab discussion**. During the Pre-Lab, you are expected to ask questions/clarification and take notes about the procedure and calculation. If you have a lot of questions, please consult the professor during office hours before the lab.

LABORATORY CONDUCT. The following guidelines apply.

- Bring your copy of the lab procedure/worksheet and scientific calculator to the lab.
- Always wear close-toed shoes. You will not be permitted to go in the lab without it.
- Short pants/skirt/dress (above the knee) and low-waist pants/skirts are not allowed unless you **wear a lab coat** over it.
- Wear safety goggles (Please purchase own. Lending goggles are very limited) as soon as you enter the lab.
- You are to work in pairs unless otherwise specified. Be alert and work cooperatively with lab partner, classmates and professor.
- Maintain a positive attitude and presence of mind. Treat the lab as an opportunity to learn the concepts and scientific process, and to acquire laboratory, teamwork and analytical skills.
- Cell phone use is strictly prohibited because it is disruptive of learning. Turn off all cell phones prior to the
 start of class. Inappropriate and disruptive behavior such as using making offensive remarks, prolonged
 chattering, reading/viewing materials not related to the course, etc. will not be tolerated. Disruptive
 students will be warned *ONCE*, and if disruptive behavior continues, this will be reported to the Security
 Office and Student Affairs.
- Obtain your materials and supplies from the laboratory cart and from your drawer.
- Follow laboratory rules and procedures at all times. Treat all chemicals with respect, replace lids on bottles and report any accident or problem to the instructor.
- Follow the directions in the Procedure precisely. Don't take short cuts nor fake results as these are readily spotted. Consult with the professor if you are not sure what to do.

- When recording measurements, the value must reflect the precision of the instrument used with the
 correct units. Never round off measurements. For example, a 10-mL graduated cylinder is always read to 2
 decimal places (e.g. 8.50 mL) whereas, a 100-mL graduated cylinder is always read to 1 decimal place (e.g.
 70.5 mL).
- Record all your data neatly in ink. Do not erase original data. If you make a mistake just put a strikethrough line.
- Do your calculations to check if results are reasonable before dismantling the setup. Repeat the
 experiment if there was a mistake. Don't falsify data to get the expected value (see cheating policy).
- Show your data and calculations to your professor and discuss the probable causes of error with the professor <u>before</u> doing the <u>repeat</u>. Don't dismantle your setup until your worksheet get the official 'signoff.'
- Use laboratory time efficiently and bear in mind that the experiment should be done at least **ten minutes** before the end of class for cleanup activity.
- When you are finished for the day, clean your glassware, dispose waste in proper containers, cap reagent bottles, and return materials, glassware, Vernier equipment and laptop computer to their proper storage areas. Clean the weighing balance and your bench-top. Please wipe down all surfaces and wash hands before leaving

GRADING LAB REPORTS.

- Point deductions will be applied to data with incorrect precision and units, and when safety, chemical transfer, cleanup and disposal procedures are not followed.
- Point deductions will be applied to missing data and missing/incorrect/incomplete responses to questions/calculations in the worksheet.
- A grading rubric will be followed when grading formal lab reports. A copy of the grading rubric and a sample formal lab report are available in Laulima.

LATE POLICY. If submitted one week after the due date, the lab report (only for labs the student was present) will be given a *grade of 70% if complete* and 65% if less than complete. The grade assigned will be *zero* if submitted much later.

MAKE-UP POLICY.

• Chemicals and supplies are available only on the day of the scheduled lab activity. Hence, **no make-up lab** is allowed. A missed lab will get zero points.

DISABILITIES ACCOMMODATIONS

If you have a physical, sensory, health, cognitive, or mental health disability that could limit your ability to fully participate in this class, you are encouraged to contact the Disability Specialist Counselor to discuss reasonable accommodations that will help you succeed in this class. Ann Lemke can be reached at 235-7448, lemke@hawaii.edu, or you may stop by Hale 'Ākoakoa 213 for more information.

TITLE IX

Title IX prohibits discrimination on the basis of sex in education programs and activities that receive federal financial assistance. Specifically, Title IX prohibits sex discrimination; sexual harassment and gender-based

harassment, including harassment based on actual or perceived sex, gender, sexual orientation, gender identity, or gender expression; sexual assault; sexual exploitation; domestic violence; dating violence; and stalking. For more information regarding your rights under Title IX, please visit: https://windward.hawaii.edu/Title IX/.

Windward Community College is committed to the pursuit of equal education. If you or someone you know has experienced sex discrimination or gender-based violence, WCC has resources to support you. To speak with someone confidentially, contact the Mental Health & Wellness Office at 808-235-7393 or Kaahu Alo, Designated Confidential Advocate for Students, at 808-235-7354 or kaahualo@hawaii.edu. To make a formal report, contact the Title IX Coordinator, Karla K. Silva-Park, at 808-235-7468 or karlas@hawaii.edu.

ACADEMIC INTEGRITY

You are allowed to <u>discuss</u> results, calculations and interpretations with your laboratory partner and classmates, but calculations and answers in the report should be completely *your own work*. Copying someone else's data or answers is cheating. Copying someone else's work from the internet, book, or publication without giving reference to the original author is plagiarism. Submitting falsified data in a subsequent lab report for a lab that you missed is dishonest. If you wish to help your classmates, you explain the topic to them but do not give your report to be copied. If your sentence/s are the same word for word with another student's, then both/all students will be assigned a grade of "F" for the activity. All cheating incidents will be reported to the Vice Chancellor for Student Affairs. To avoid copying answers from old labs, you are <u>required</u> to turn in all your lab reports to the professor at the end of the semester.

ALTERNATE CONTACT INFORMATION

If you are unable to contact the instructor, have questions that your instructor cannot answer, or for any other issues, please contact the Academic Affairs Office:

Location: Alakai 121 Phone: 808-235-7422 Email: wccaa@hawaii.edu

TENTAIVE COURSE SCHEDULE

Week	Day	Experiment Title
1	Monday, 8/21	1- Laboratory Safety, Equipment & Procedures
2	Monday, 8/28	2- Scientific Measurements & Density
3	Monday, 9/4	Labor Day (No Lab)
4	Monday, 9/11	3- Separation of Mixtures
5	Monday, 9/18	4- Conductivity of Solutions
6	Monday, 9/25	5- Determination of Chemical Formula
7	Monday, 10/2	6- Types of Chemical Reactions
8	Monday, 10/9	7- Beer's law investigation **
9	Monday,10/16	8- Titration of acetic acid in vinegar
10	Monday, 10/23	9- Stoichiometry Experiment
11	Monday, 10/30	10 - Calorimetry and Hess's law
12	Monday, 11/6	11- Periodic trends
13	Monday, 11/13	12- Lewis structures and molecular shapes
14	Monday, 11/20	Thanksgiving Week
15	Monday, 11/27	13- Spectroscopy and identification of Halide
16	Monday, 12/4	Lab Final
17	Monday, 12/11	Finals week for lectures

^{**} Requires formal Lab report