

Chem 100L Chemistry in Society Laboratory

1 credit (CRN 63117)

Friday 10 am - 12:45 pm

Imiloa 111 & 131

INSTRUCTOR: Leticia Colmenares, Ph.D.
OFFICE: Imiloa 116
E-MAIL: Leticia@hawaii.edu
OFFICE HOURS: MW 10:00-11:00 am
TR 11:30-12:30 pm
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EFFECTIVE DATE: Fall 2016

WINDWARD COMMUNITY COLLEGE MISSION STATEMENT

Windward Community College is committed to excellence in the liberal arts and career development; we support and challenge individuals to develop skills, fulfill their potential, enrich their lives, and become contributing, culturally aware members of our community.

CATALOG DESCRIPTION

Experiments in everyday chemistry. (2 hrs. 45 min lab.)

Prerequisites: Credit or registration in CHEM 100.

WCC: DY

STUDENT LEARNING OUTCOMES

1. Identify/locate laboratory safety equipment and apply laboratory safety procedures.
2. Construct molecular models to determine molecular shape and properties.
3. Assemble apparatus to perform common laboratory techniques to verify fundamental chemistry principles in everyday life.
4. Make and record accurate observations and precise quantitative measurements.
5. Synthesize conclusions based on observations and data in a formal laboratory report.
6. Identify sources of error in laboratory experiments.

LEARNING RESOURCES

Required Textbook: Chemistry 100L Laboratory Manual Spring 2016, edited by L. Colmenares

Course Website: <http://laulima.hawaii.edu> (use UH email account login and password)

Other Requirements: Chem 100 Lecture Notes/textbook, Calculator, Internet Access, Lab goggles, closed shoes and a lab gown if you wear short pants/skirt/dress or low-waist pants/skirts.

PURPOSE OF THE LABORATORY COURSE

The chemistry laboratory allows the student to understand some of the concepts discussed in the lecture more thoroughly. In the laboratory you will be involved with the processes of scientific inquiry. It is the only way for the student to learn the techniques that are so important in chemistry and applied sciences. The student will discover that doing quality work in the laboratory requires a great deal of patience and care.

GRADING

The final grade will be based on the following scheme:

10 highest (out of 11) Lab Reports*	70% of total
Midterm	15% of total
Final Exam	15% of total
Total	100%

*The lowest lab report score will be deleted.

Course grades will be assigned as follows:

A-- 90-100%

B-- 80-89%

C-- 70-79%

D-- 60-69%

F-- less than 60%.

For grades I, W, Cr, NC-- See college catalog,

https://windward.hawaii.edu/Catalogs_Schedules/WCC_Catalog_current.pdf

MODE OF INSTRUCTION

The primary mode of instruction is through the use of hands-on laboratory activities. Each hands-on activity will be preceded by a class discussion based on a pre-lab online homework consisting of a reading assignment and completion of pre-lab questions and problems. The homework and pre-lab may consist of videos, websites and computer activities.

Conduct of the Lab:

First 45 min	Discuss answers to <u>pre-lab</u> assignment problems and questions, review important principles and safety precautions, demonstrate new techniques needed or use of new equipment (in Imiloa 111).
Remaining time	Perform hands-on activity, record observation and data, calculate results (in Imiloa 131)
10 min	Clean-up
Last 30 minutes	Answer Post-lab questions

Before leaving the laboratory, you must have your **lab report signed and dated by the instructor**. The completed lab report may be turned in at the end of the class, or at the start of next meeting (final due date). Only lab reports that have signature will be accepted.

COURSE POLICIES AND TIPS TO SUCCEED IN THIS COURSE

1. The **Course Content and Schedule** (activity titles and corresponding dates) is listed on the last page. Changes in the schedule will be announced in class and course website. Use this syllabus as your course guide throughout the semester. You are expected to login to the Laulima course website at least once a week.
2. A lab activity is scheduled for each meeting. You are expected to turn in a lab report for each activity. The maximum points for each activity is **20 points**.
3. **Before class.** Please allot 2-3 hours weekly outside class time: (1) study the lab manual and videos/resources in Laulima Modules (2) take the pre-lab quiz in Laulima (3) complete your lab report that is due.
You are expected to become familiar with the concept, procedure and calculations that will be done in the lab.

Take the **online pre-lab quiz** before coming to class. The online quiz is available continuously for one week. It closes 30 minutes before the class starts. Your quiz score constitutes **2 points**.

4. Bring your **Lab Manual, scientific calculator, Lecture Notes** and **a stapler** to class at all times.
5. **Safety** is important. Wear closed-toed shoes (preferably sports shoes) to protect from danger of spilling chemicals. Slippers, sandals or similar footwear are not allowed. **You will NOT be permitted in the lab** if you are not wearing close-toed shoes. Short pants/skirt/dress (above the knee) and low-waist pants/skirts are not allowed unless you wear a lab coat over it.
6. **Pre-Lab period (Imiloa 111).** Please come to class on time. Important background information about the experiment and safety will be discussed in the Pre-Lab discussion. During the Pre-Lab, you are expected to ask questions/clarification about the procedure and calculation. If you have a lot of questions, please see the instructor during office hours. Please take notes during the pre-lab session.

Attendance will be checked at the beginning of each Pre-Lab session. If you arrive late, you will be deducted 0.5 points from your lab report score. If you miss the pre-lab session, you will be deducted 1 point from your lab report score.

Extra Credit: It is highly recommended to have a separate laboratory notebook. If you take notes and dates, record data, observations and calculations for each experiment in a **separate notebook** and turn in the laboratory notebook at the end of the semester, you will get an extra credit of 5 points (total).

Other extra credit opportunities: If you participate in the WCC Hoolaulea on Oct 1, 2016, you will get 10 points. If you attend any chemistry forum, you will get 2 points.

7. **During lab (Imiloa 131). Wear safety goggles (found in lab cabinet) as soon as you enter the lab.** Follow the **directions in the Procedure** precisely. Gather your materials and supplies from the laboratory cart and from your drawer. Don't take short cuts nor fake results as these are readily spotted. Follow **laboratory rules and procedures** at all times. Treat all chemicals with respect, replace lids on bottles and report any accident or problem to the instructor. Points will be **deducted** when safety and chemical transfer procedures are not followed.
8. In the lab, you are to work in pairs. You will have a different partner each week. This will be randomly assigned to you. Work cooperatively and maintain a positive attitude. Treat the lab as an **opportunity to learn teamwork**. Do **not** simply rush through an experiment in order to get out of the laboratory. **Maintain a positive attitude** and **work cooperatively** with other students and the laboratory instructor. **Be alert** and maintain **presence of mind**.
9. **Inappropriate and disruptive behavior** such as using cell phones, making offensive remarks, prolonged chattering, reading/viewing materials not related to the course, etc. will not be tolerated. Disruptive students will be warned **ONCE**, and if disruptive behavior continues, this will be reported to the Security Office and Student Affairs.
10. When recording **measurements**. The **value must** reflect the precision of the instrument used. **Never round off measurements**. Affix the correct unit. **Record all your data** neatly in ink and with units. Do not erase original data. If you make a mistake just put a strikethrough line. Point deductions will be applied to data with incorrect precision and units.
11. Do your calculations to check if results are reasonable before dismantling the setup. Show your calculation to the instructor. **Repeat the experiment** if there was a mistake. **Discuss** the probable **causes of error with the instructor** before doing the repeat.
12. Use laboratory time efficiently and bear in mind that the experiment **should be done** at least **ten minutes** before the end of class for cleanup activity. When you are finished for the day, clean your glassware, dispose waste in proper containers, cap reagent bottles, and return materials, glassware, Vernier equipment and laptop computer to their proper storage areas. Clean the weighing balance and your bench-top. Points will be **deducted** when clean up and waste disposal techniques and procedures are not followed.

A class monitor team will be assigned to supervise, check and report the cleanup. The team will take care of neutralizing waste solutions and will be assigned on a rotation.
13. **After completing the lab**. Show your data and responses to your instructor and request for her signature.
14. **Laboratory reports**. You are expected to turn in an **individual** report for each lab experiment at the start of the next meeting. Please include all data and calculations, and answer ALL questions. Use **internet resources** and **Lecture Notes/textbook** to answer post-lab questions. Points will be deducted for omissions or incorrect

answers.

You are expected to discuss results, calculations and interpretations with your laboratory partner and classmates, but calculations and answers in the report should be completely ***your own work***.

Cheating policy. Copying someone else's data or answers is cheating. If your sentence/s are the same word for word with another student's, then both students will be assigned a grade of "F" for the activity and will be reported to the Vice Chancellor of Student Affairs. If you copy someone else's work from the internet/book/publication without giving reference to the original author, this is plagiarism and an act of dishonesty. Also, you should NOT turn in a lab report for the activity you were absent.

15. A ***formal*** laboratory report is required for Experiments on *Polymers* and *Household Chemicals* while an ***informal*** lab report is required for all the other experiments. ***Formal reports*** should be typewritten and follow the standard format. The ***rubric for grading*** the formal lab reports and tips on ***how to write a formal report*** are found in your Lab Manual. A sample formal lab report is downloadable at Laulima course website.
16. **Late Reports and Penalties.** If submitted one week after the due date, the lab report (only for labs the student was present) will be given a ***grade of 70% if complete*** and 60% if less than complete. The grade assigned will be ***zero*** if submitted much later.
17. Each **lab report** is worth 18-20 points. You will get your ***graded lab report*** at the following lab session. However, you are to return it to the instructor immediately after reviewing it. The instructor will keep the graded lab reports but you will be allowed to take them home the week before the exam. You are to return these to the instructor on the day of the exam.

Monitor your class performance by keeping a record of your scores on the table of contents page of the Lab Manual. Check your grades in Laulima Gradebook.
18. **Make-up.** Chemicals and supplies are available only on the day of the lab activity. Hence, no make-up is allowed and the student gets "zero" for the missed lab activity. You cannot turn in a lab report for a missed activity.
19. If you have any **special learning needs**, including hearing/visual impairment, please inform the instructor as soon as possible.
20. Please communicate with the instructor if you can't attend class as soon as possible.

DISABILITIES ACCOMMODATION

If you have a physical, sensory, health, cognitive, or mental health disability that could limit your ability to fully participate in this class, you are encouraged to contact the

Disability Specialist Counselor to discuss reasonable accommodations that will help you succeed in this class. Ann Lemke can be reached at 235-7448, lemke@hawaii.edu, or you may stop by Hale 'Akoakoa 213 for more information.

COURSE CONTENT AND SCHEDULE

Important Dates: Non-Instructional Sep 5 (M), Nov 8 (R), 11 (F), 24-25 (R-F)

Last day for withdrawal and change grade option, Nov 3 (R)

Last day of instruction, Dec 8 (R)

Date	Activity	Notes
Aug 26	1- Laboratory Safety, Equipment & Procedures	
Sep 2	2- The Scientific Method	
Sep 9	3- Elements and Compounds	
Sep 16	4- Measurements & Density	
Sep 23	5- Chromatography	
Sep 30	7- Stoichiometry of a Precipitation Reaction	
Oct 7	8- Acids, Bases & Neutralization	
Oct 14	Midterm Exam (Experiment 1-7)	How to Write a Formal Lab Report
Oct 21	<i>10- Antacids and Sunscreens</i>	
Oct 28	9- Reduction-Oxidation or Food Chemistry (handout)	
Nov 4	11- <i>Polymers</i> **	Formal Lab Report
Nov 11	Veterans Day	
Nov 18	<i>12- Household Chemicals</i> **	Formal Lab Report
Nov 25	Holiday	
Dec 2	Review & Final Long Exam (Experiment 8-12)	

***Subject to change**

**** Require formal lab reports-see sample in Laulima**